

STUDY MODULE DESCRIPTION FORM		
Name of the module/subject Fundamentals of Electrochemical Technology		Code
Field of study Chemical and Process Engineering	Profile of study (general academic, practical) general academic	Year /Semester 3/6
Elective path/specialty	Subject offered in: Polish	Course (compulsory, elective) elective
Cycle of study: first-cycle studies	Form of study (full-time, part-time) full-time	
No. of hours Lecture: - Classes: - Laboratory: 1 Project/seminars: -		No. of credits 1
Status of the course in the study program (Basic, major, other) Basic		(university-wide, from another field) University-wide
Education areas and fields of science and art Technical science		ECTS distribution (number and %) 1 100 %
Responsible for subject / lecturer: Prof. dr hab. Jan Skowroński tel. 616653641 Wydział Technologii Chemicznej ul. Piotrowo 3 60-965 Poznań		
Prerequisites in terms of knowledge, skills and social competencies:		
1	Knowledge	Ordered knowledge of mathematics and physical chemistry.
2	Skills	Ability to use the basic techniques in a laboratory scale.
3	Social competencies	The need for further education and enhance of professional and personal competences.
Assumptions and objectives of the course: The aim of the course is to broaden the knowledge as well as reinforcing the skills to plan and conduct electrochemical processes used in practice.		
Study outcomes and reference to the educational results for a field of study		
Knowledge: 1. The knowledge in the field of basics of electrochemical processes –[K_W03, K_W04], 2. The knowledge in the field of various electrochemical technologies –[K_W13, K_W15], 3. The knowledge in the field of related fields –[K_W12].		
Skills: 1. The student can use in practice theoretical knowledge gained earlier –[K_U08, K_U15, K_U16], 2. The student has the ability to selection of measurement techniques –[K_U01, K_U02],		
Social competencies: 1. The student understands the need for self-study and improvement of their professional competence –[K_K01], 2. Student can act and cooperate in the group accepting different roles –[K_K04].		
Assessment methods of study outcomes		
Laboratory assessment on the basis of the current work during the laboratory and the written tests.		
Course description		

1. Electrode materials used in electrochemical technologies. 2. Electrochemical techniques used in practice in electrochemical processes. 3. The examples of electrochemical synthesis.		
Basic bibliography: 1. A. Kiswa – Elektrochemia cz. I i II (Jonika i Elektrodyka) WNT, W-wa, 2001, 2. R. Dylewski, W. Gniot, M. Gonet, Elektrochemia przemysłowa, Wyd. Politechniki Śląskiej, 1999, 3. A. Czerwiński, "Ogniwa, akumulatory, baterie", WNT, W-wa, 1999.		
Additional bibliography: 1. A.V. da Rosa, „Fundamentals of Renewable Energy Processes” Elsevier/Academic Press, 1990, 2. H. Scholl, T. Błaszczyk, P. Krzyczmonik, Elektrochemia, Wyd. Uniwersytetu Łódzkiego, 1998.		
Result of average student's workload		
Activity	Time (working hours)	
1. consultation to the laboratory	2	
2. preparation for the laboratory	4	
3. laboratory	15	
Student's workload		
Source of workload	hours	ECTS
Total workload	21	1
Contact hours	17	1
Practical activities	4	0